

FLAVONOIDS FROM *Gossypium hirsutum* FLOWERS

T. Wu,^{1,2} R. Abdulla,^{1,2} Y. Yang,^{1,2} and H. A. Aisa^{1*}

UDC 547.972

Gossypium hirsutum L. (Malvaceae) is a cotton species that is widely cultivated worldwide. In Xinjiang Autonomous Region (PRC), flowers of *G. hirsutum* are used in folk medicine. The plant is officially listed as a Chinese Medicinal Herb and is used as a sedative agent and to treat delayed mental development and to improve brain function [1]. We studied the flavonoid composition of flowers of *G. hirsutum*.

Flowers of *G. hirsutum* (2 kg) were extracted with ethanol (95%, 3 times). The extracts were combined and evaporated to dryness at reduced pressure to afford dry powder (400 g) that was fractionated over a column of ion-exchange resin AB-8 (weakly polar). Compounds were eluted from the column by water and aqueous ethanol (50 and 70%). The alcohol fractions (90 g) were rechromatographed over a column of silica gel to produce **1-7**.

Kaempferol-3-O-β-D-(6"-O-p-coumaryl)glycoside (1), light-yellow powder, mp 246-248°C (MeOH). UV spectrum (λ_{\max} , MeOH, nm): 226, 268, 316, 352sh.

PMR spectrum (600 MHz, DMSO-d₆, δ, ppm, J/Hz): 5.46 (1H, d, J = 7.8, H-1''), 6.12 (1H, d, J = 16.2, H-8'''), 6.16 (1H, d, J = 1.8, H-6), 6.39 (1H, d, J = 1.8, H-8), 6.79 (2H, d, J = 8.4, H-3'',5''), 6.86 (2H, d, J = 9.0, H-3',5'), 7.35 (1H, d, J = 16.2, H-7'''), 7.38 (2H, d, J = 8.4, H-2'',6''), 8.00 (2H, d, J = 9.0, H-2',6'), 10.03, 10.17 (1H each, br.s, OH on C-4'' and C-4', respectively), 10.86 (1H, br.s, 7-OH), 12.59 (1H, s, 5-OH).

¹³C NMR spectrum (150 MHz, DMSO-d₆, δ, ppm): 177.6 (C-4), 166.4 (C-9'''), 164.4 (C-7), 161.4 (C-5), 160.2 (C-4'), 160.0 (C-4'''), 156.6 (C-2), 156.5 (C-9), 144.8 (C-7'''), 133.2 (C-3), 131.0 (C-2',6'), 130.4 (C-2'',6''), 125.1 (C-1'''), 121.0 (C-1'), 116.0 (C-3',5'), 115.3 (C-3'',5''), 113.8 (C-8'''), 104.1 (C-10), 101.1 (C-1'), 99.0 (C-6), 93.9 (C-8), 76.4 (C-3'), 74.4 (C-5''), 74.3 (C-2''), 70.1 (C-4''), 63.2 (C-6'') [2].

Quercetin (2), yellow powder, mp >300°C (acetone). UV spectrum (λ_{\max} , MeOH, nm): 256, 374.

PMR spectrum (600 MHz, DMSO-d₆, δ, ppm, J/Hz): 6.19 (1H, d, J = 1.8, H-6), 6.41 (1H, d, J = 2.4, H-8), 6.88 (1H, d, J = 8.4, H-5'), 7.55 (1H, q, J = 8.4, 2.4, H-6'), 7.68 (1H, d, J = 2.4, H-2'), 9.32 (1H, s, 4'-OH), 9.39 (1H, s, 3-OH), 9.61 (1H, s, 3'-OH), 10.79 (1H, s, 7-OH), 12.50 (1H, s, 5-OH).

¹³C NMR spectrum (150 MHz, DMSO-d₆, δ, ppm): 93.55 (C-8), 98.38 (C-6), 103.21 (C-10), 115.80 (C-2'), 116.39 (C-5'), 120.17 (C-6'), 122.15 (C-1'), 135.95 (C-3), 145.26 (C-3'), 146.99 (C-2), 147.91 (C-4'), 156.32 (C-9), 160.92 (C-5), 164.09 (C-7), 176.05 (C-4) [3].

Quercetin-3'-O-β-D-glucoside (3), yellow powder, mp 183-184°C (MeOH). UV spectrum (λ_{\max} , MeOH, nm): 252, 370.

PMR spectrum (600 MHz, DMSO-d₆, δ, ppm, J/Hz): 3.17-5.09 (sugar protons), 5.15 (1H, d, J = 4.2, H-1''), 6.19 (1H, d, J = 1.8, H-6), 6.49 (1H, d, J = 1.8, H-8), 6.98 (1H, d, J = 8.4, H-5'), 7.85 (1H, q, J = 8.4, 2.4, H-6'), 7.96 (1H, d, J = 1.8, H-2'), 9.22 (1H, s, 4'-OH), 9.42 (2H, s, 3-OH), 10.67 (1H, s, 7-OH), 12.46 (1H, s, 5-OH).

¹³C NMR spectrum (150 MHz, DMSO-d₆, δ, ppm): 60.76 (C-6''), 69.73 (C-4''), 73.49 (C-2''), 76.14 (C-3''), 77.42 (C-5''), 93.90 (C-8), 98.44 (C-6), 102.59 (C-1') 103.22 (C-10), 115.93 (C-2'), 116.20 (C-5'), 122.42 (C-6'), 123.77 (C-1'), 136.20 (C-3), 145.49 (C-3'), 146.43 (C-2), 148.98 (C-4'), 156.39 (C-9), 160.84 (C-5), 164.22 (C-7), 176.13 (C-4) [4, 5].

Quercetin-3-O-β-D-glucoside (4), yellow powder, mp 249-250°C (MeOH). UV spectrum (λ_{\max} , MeOH, nm): 257, 361.

1) Xinjiang Technical Institute of Physics and Chemistry, Academy of Sciences of the PRC, Urumchi, 830011, China, fax (86991) 383 56 79, e-mail: haji@ms.xjb.ac.cn; 2) Graduate School of the PRC, Beijing, 100039, PRC. Translated from Khimiya Prirodnnykh Soedinenii, No. 3, pp. 296-297, May-June, 2008. Original article submitted April 3, 2008.

PMR spectrum (600 MHz, DMSO-d₆, δ, ppm, J/Hz): 3.07-5.29 (sugar protons), 5.47 (1H, d, J = 7.2, H-1''), 6.20 (1H, d, J = 2.4, H-6), 6.40 (1H, d, J = 1.8, H-8), 6.84 (1H, d, J = 8.4, H-5'), 7.58 (1H, d, J = 2.4, H-6'), 7.56 (1H, br.s, H-2'), 9.24 (1H, s, 4'-OH), 9.74 (1H, s, 3'-OH), 10.86 (1H, s, 7-OH), 12.65 (1H, s, 5-OH).

¹³C NMR spectrum (150 MHz, DMSO-d₆, δ, ppm): 61.97 (C-6''), 70.13 (C-4''), 74.29 (C-2''), 76.69 (C-3''), 77.81 (C-5''), 91.02 (C-8), 98.85 (C-6), 101.01 (C-1''), 104.18 (C-10), 115.40 (C-2'), 116.39 (C-5'), 121.37 (C-1'), 121.82 (C-6'), 133.50 (C-3), 145.02 (C-3'), 148.66 (C-4'), 156.36 (C-9), 156.52 (C-2), 161.45 (C-5), 164.32 (C-7), 177.65 (C-4) [6].

Hyperoside (5), yellow powder, mp 234-236°C (MeOH). UV spectrum (λ_{max} , MeOH, nm): 257, 305sh, 361.

PMR and ¹³C NMR spectra of **5** agreed fully with those published [3]. Acid hydrolysis of **5** produced quercetin and D-galactose [6].

Astragalin (6), yellow needles (MeOH). UV spectrum (λ_{max} , MeOH, nm): 266, 300sh, 351.

PMR and ¹³C NMR spectra of **6** agreed fully with those published [3]. Acid hydrolysis of **6** gave kaempferol and D-glucose [7].

Quercetin-7-O-β-D-glucoside (7), yellow powder, mp 210-212°C (MeOH). UV spectrum (λ_{max} , MeOH, nm): 256, 373.

PMR spectrum (400 MHz, DMSO-d₆, δ, ppm, J/Hz): 3.20-3.72 (sugar protons), 5.07 (1H, d, J = 7.2, H-1'), 6.43 (1H, d, J = 2, H-6), 6.78 (1H, d, J = 2, H-8), 6.94 (1H, d, J = 8.4, H-5'), 7.58 (1H, d, J = 8.4, 2, H-6'), 7.72 (1H, d, J = 2, H-2'), 9.39 (1H, s, 4'-OH), 9.55 (1H, s, 3-OH), 9.76 (1H, s, 3'-OH), 12.49 (1H, s, 5-OH).

¹³C NMR spectrum (100 MHz, DMSO-d₆, δ, ppm): 61.11 (C-6''), 70.04 (C-4''), 73.60 (C-2''), 76.78 (C-3''), 77.69 (C-5''), 94.99 (C-8), 99.35 (C-6), 100.47 (C-1''), 105.23 (C-10), 115.87 (C-2'), 116.16 (C-5'), 120.76 (C-6'), 122.44 (C-1'), 136.59 (C-3), 145.50 (C-3'), 148.08 (C-2), 148.31 (C-4'), 156.38 (C-9), 160.70 (C-5), 163.28 (C-7), 176.51 (C-4) [5].

REFERENCES

1. *China Pharmacopoeia Committee, Materia Medica Criterion, the Division of Uigur Materia Medica*, Beijing (1999), p. 98.
2. J. G. Luo, L. Y. Kong, Z. Guo, and Z. Yao, *Za Zhi*, **30**, 516 (2005).
3. Z. P. Xiao, H. K. Wu, T. Wu, and H. A. Aisa, *Chem. Nat. Comp.*, **42**, 736 (2006).
4. T. A. Geissman, *The Chemistry of Flavonoid Compounds*, Pergamon Press, Oxford (1962), 108.
5. K. R. Markham, B. Ternai, R. Stanley, H. Geiger, and T. J. Mabry, *Tetrahedron*, **34**, 1389 (1978).
6. G. Ye and C. G. Huang, *Chem. Nat. Comp.*, **42**, 232 (2006).
7. B. H. Hu, Z. Tian, and Z. C. Lou, *Acta Bot. Sin.*, **30**, 565 (1988).